

NICKEL-CATALYZED REGIO- AND STEREO-SELECTIVE SYNTHESIS OF HOMOALLYLIC ALCOHOL DERIVATIVES FROM DIENES AND ALDEHYDES

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Abstract: The reaction of 1,3-Dienes and aldehydes in the presence of Et_3SiH using a catalytic amount of $\text{Ni}(\text{cod})_2$ and PPh_3 gave homoallylic alcohol derivatives in a regio- and stereoselective manner.

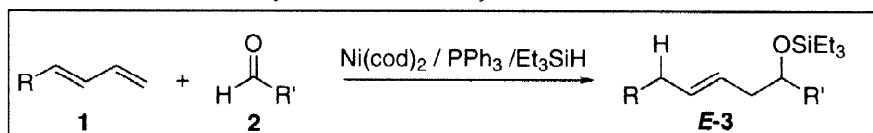
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Many reports concerning nickel-catalyzed reaction of 1,3-dienes and multiple bonds have indicated that these processes are effective for carbon-carbon bond formations¹⁻⁴. However, it is difficult to form a new carbon-carbon bond intermolecularly from two different multiple bonds because of their low selectivities^{5,6}. Here, we report a selective intermolecular coupling reaction of 1,3-diene and aldehyde.

In the course of studies on nickel-catalyzed intramolecular diene-aldehyde cyclization, we found that the reaction of 1,3-diene having an aldehyde moiety in a chain in the presence of Et_3SiH proceeded by a catalytic amount of zero-valent nickel complex in a highly regio- and stereoselective manner⁷. Based on these results, it was thought that this intramolecular coupling reaction might be also applicable to highly regio- and stereo-controlled synthesis of homoallylic alcohol derivatives from 1,3-diene and aldehyde (**Scheme 1**).

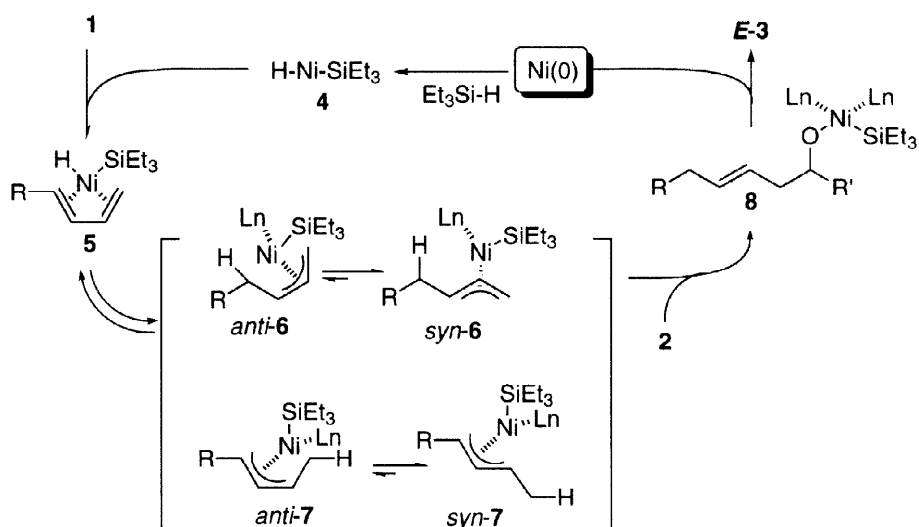
Scheme 1 Plan for the synthesis of homoallylic alcohol derivatives



Our plan is outlined in **Scheme 2**. In this reaction, nickel hydride complex **4**, generated by the oxidative addition of Et_3SiH to zero-valent nickel complex, is coordinated by diene **1** to give complex **5** (**Scheme 2**). The insertion of the diene moiety of **5** into the nickel-hydride bond produces π -allylnickel complexes *anti*-**6** and *anti*-**7**, which would be in a state of

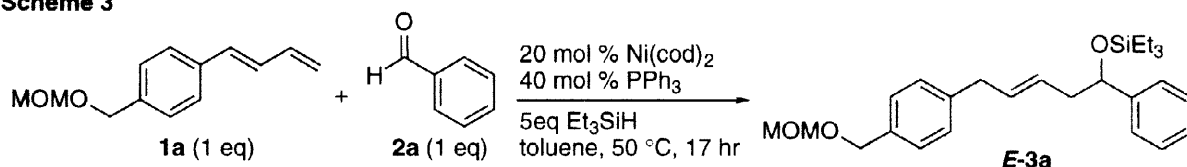
equilibrium with the more stable *syn*-6 and *syn*-7.⁸ Then, *syn*-6 would react with aldehyde 2 at the less substituted terminal carbon of the complex *syn*-6 to provide complex 8. Reductive elimination from complex 8 should give silyl-protected homoallyl alcohol derivative *E*-3, and zero-valent nickel complex would be regenerated.

Scheme 2



To examine the feasibility of the above plan, the coupling reaction of diene **1a** and benzaldehyde **2a** was carried out (Scheme 3). When a toluene solution of diene **1a** (1 eq), aldehyde **2a** (1 eq), and Et₃SiH (5 eq) was stirred in the presence of Ni(cod)₂ (20 mol %) and PPh₃ (40 mol %) at 50 °C for 17 hr, triethylsilyl-protected alcohol *E*-3a was obtained in 84% yield as a sole product in a regio- and stereoselective manner.

Scheme 3



Encouraged by this result, further investigation of coupling reactions between dienes **1** and various aldehydes **2** were carried out. The results are shown in Table 1. The reaction of diene **1a** with aldehyde **2b**, having an electron-donating group at the 4-position on the aromatic ring, gave the desired product in 76% yield (entry 1). In a similar manner, the reaction with **2c**, having an electron-withdrawing group at the same position, also provided a coupling product in moderate yield along with the recovery of diene **1a** in 32% yield (entry 2). In the reactions with aliphatic aldehydes **2d** and **2e**, coupling products were obtained in good yields as a mixture of *E*- and *Z*-isomers. It was interesting that the ratio of *E*- to *Z*-isomers was improved when the initial concentration of the reagents was changed (entries 4 and 5). The reaction of **1b** with various aldehydes (entries 6~9) also afforded coupling products in a regio-

and stereoselective manner in moderate yields. In these reactions, a fair amount of hydrosilylation product **9** was obtained (entries 6, 7, and 9).

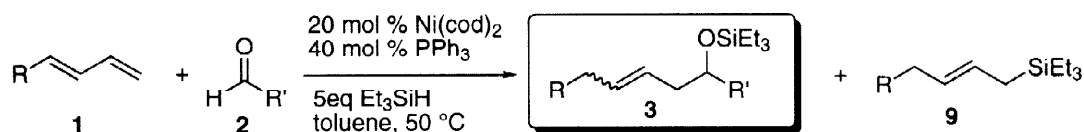


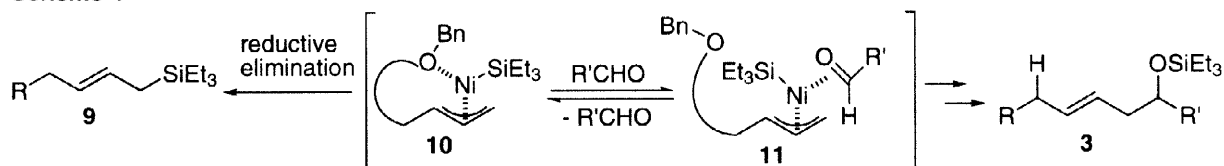
Table 1. Reaction of Various Dienes and Aldehydes in the presence of Et₃SiH and Nickel Catalyst.

entry ^a	diene (R =)	aldehyde (R' =)	time (hr)	yield of 3 (%)	ratio (E- 3 / Z- 3)	yield of 9 (%)
1			18	76	100 / 0	0
2	1a		44	59 ^c	100 / 0	0
3	1a		27	71	3 / 1	0
4	1a		23	76	1 / 1	0
5 ^b	1a	2e	24	81	2 / 1	0
6			18	63	100 / 0	7
7	1b	2b	19	49	100 / 0	31
8	1b	2c	44	54 ^d	100 / 0	0
9	1b	2e	18	63	100 / 0	11
10		2a	20	72	100 / 0	0
11	1c	2b	19	66	100 / 0	0

^a The initial concentrations of diene and aldehyde were 0.05 M except for entry 5. ^b The initial concentrations of diene and aldehyde were 0.02 M. ^c 32% of **1a** was recovered. ^d 35% of **1b** was recovered.

It seemed likely that the reactivity of π -allylnickel complex generated from diene **1b** was lower than that generated from **1a**. The lower reactivity can be explained by the coordination of oxygen atom of the benzyloxy group onto the nickel center of **10** (Scheme 4). Thus, aldehyde could not coordinate to nickel center, and reductive elimination from **10** would occur to give hydrosilylated product **10**.

Scheme 4



To confirm this, coupling reactions of diene **1c** were carried out. In the reaction with aldehyde **2a**, a coupling product was obtained in 72% yield as a sole product (entry 10), and the reaction with aldehyde **2b** also gave only a coupling product in 66% yield (entry 11). These results indicate that the coupling reactions of **1b** were disturbed by coordination of the benzyloxy group onto the nickel center.

In conclusion, we developed a highly regio- and stereoselective synthesis of homoallyl alcohol derivatives by the intermolecular coupling reaction of 1,3-dienes and aldehydes using a catalytic amount of nickel complex in the presence of Et₃SiH. Further studies along this line are in progress.

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- [8] Hydrosilylation of 1,3-pentadiene using Ni(acac)₂-Et₃Al was reported by Lappert. In this case, 1,3-pentadienes gave principally 2-pentenylsilane derivatives. They claimed that secondary alkyl-nickel complex is formed at first and then it rearranged to primary-nickel complex. However, it was considered that these π -allylnickel complexes are in a state of equilibrium and *syn*-**6** should react with aldehyde because *syn*- π -allylnickel complex is more stable than *anti*-complex and the reactivity of *syn*-**6** would be higher than that of *syn*-**7**. See Lappert, M. F.; Nile, T. A.; Takahashi, S. *J. Organomet. Chem.* **1974**, *72*, 425.